

CH₃CN complex with water as a precursor for radiation induced synthesis of acetamide in low-temperature matrix

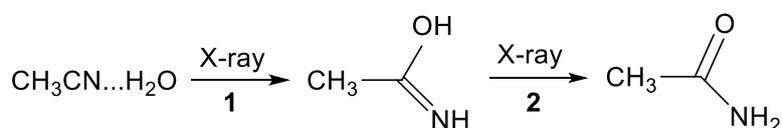
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Small nitrogen-containing molecules play an important role in extraterrestrial prebiotic chemistry. In particular, acetonitrile (CH₃CN) was found in different space objects.¹ It is supposed to be a precursor of a variety of biologically relevant molecules, including amino acids². Considering the solid-phase chemistry driven by ionizing radiation, one may come to an idea that the intermolecular complexes of acetonitrile with oxygen-bearing molecules can serve as “building blocks” for complex organic molecules (COMs) containing both N and O atoms. However, little is known on the mechanisms of such processes. The strategy to investigation of the radiation-chemical transformations occurring within isolated complexes frozen in rigid environment was recently developed in our laboratory.^{3,4} In this work we have examined for the first time the radiation-induced transformations of acetonitrile complexes with the most common space molecules (water and carbon dioxide) occurring under X-ray irradiation in solid argon matrices at 5 – 6 K using FTIR spectroscopy.

As a first step, the 1 : 1 complex of acetonitrile with water in two conformations was experimentally obtained by condensation of the ternary gaseous mixtures (CH₃CN/H₂O/Ar) and characterized on the basis of comparison with available computational data.⁵ Then, it was shown that the radiolysis of complexes with X-rays led to formation of oxygen-containing COMs, such as acetamide, acetimidic acid, hydroxyacetonitrile and acetonitrile N-oxide. The formation of these products was also confirmed by experiments with isotopic substitution (CD₃CN). Remarkably, the built-up kinetic curves demonstrated induction period for acetamide and maximum for acetimidic acid, which may indicate a new two-step route to the cold radiation-induced synthesis of this important prebiotic molecule in the space environment. Comparison between direct and reverse process shown earlier⁶ is discussed and astrochemical implications of the results are outlined.



Scheme 1. Formation of acetamide via radiation-induced assemble in acetonitrile-water complex (1) and intermolecular rearrangement (2)

1. Cottin, H. & Fray, N. Distributed sources in comets. *Space Sci. Rev.* **138**, 179–197 (2008).
2. Hudson, R. L., Moore, M. H., Dworkin, J. P., Martin, M. P. & Pozun, Z. D. Amino acids from ion-irradiated nitrile-containing ices. *Astrobiology* **8**, 771–779 (2008).
3. Zaslomov, P. V. Radiation-induced chemistry in the C₂H₂-H₂O system at cryogenic temperatures: a matrix isolation study. *Mon. Not. R. Astron. Soc.* **491**, 5140–5150 (2020).
4. Lukianova, M. A. *et al.* Radiation-induced transformations of HCN··C₂H₂, HCN··C₂H₄ and HCN··C₂H₆ complexes in noble gas matrices: Synthesis of C₃H_xN molecules in cryogenic media. *Radiat. Phys. Chem.* **180**, 109232 (2021).
5. Gopi, R., Ramanathan, N. & Sundararajan, K. Acetonitrile-water hydrogen-bonded interaction: Matrix-isolation infrared and ab initio computation. *J. Mol. Struct.* **1094**, 118–129 (2015).
6. Duvernay, F., Chatron-Michaud, P., Borget, F., Birney, D. M. & Chiavassa, T. Photochemical dehydration of acetamide in a cryogenic matrix. *Phys. Chem. Chem. Phys.* **9**, 1099–1106 (2007).

Acknowledgments: This work was financially supported by the Russian Science Foundation (project no. 21–13–00195).