

Molecular IR Emission Spectra of Solid C₆₀ and C₇₀ Fullerenes

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Cosmic abundance of fullerene molecules has been a focus in astrophysics and interstellar chemistry, since the detection of the infrared (IR) emission bands of C₆₀ and C₇₀ in the planetary nebula.¹ Concerning IR absorption spectra, vibrational fingerprints of C₆₀ in solid pH₂ were understood by the presence of a number of isotopomers, ¹³C_{x¹²C_{60-x} ($x = 0-3$), and discussed along the Einstein's B coefficient.² For the estimation of molecular abundance in space, vibrational temperature is crucial, because the IR emission requires population of molecules in their vibrationally excited states.}

We measured IR emission spectra of a thin plate of solid C₆₀ in the laboratory for the study of temperature dependence of the intensity ratio of the four IR-active vibrational T_{1u} modes of the icosahedral molecule in a range of 300–370 K.³ Figure 1 shows typical IR emission spectra of C₆₀ and C₇₀ at 353 K (80°C). The four IR emission bands of C₆₀ easily saturate to the blackbody contour and numerous combination bands and overtones are intensified in-between. The same characteristics apply to the observed IR emission spectra of C₇₀. A relatively simple model based on Boltzmann distribution for the $v = 1$ level of normal modes in the individual molecule is compatible with the observed intensity ratios of the four IR emission bands of C₆₀. Figure 2 depicts simulated temperature dependence of the IR emission intensity for the T_{1u} modes of C₆₀.

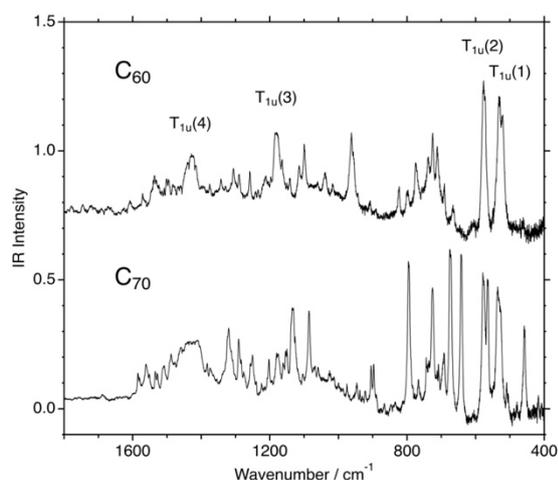


Fig. 1. IR emission spectra of thin plates of C₆₀ and C₇₀ fullerenes at elevated temperature of 353 K.

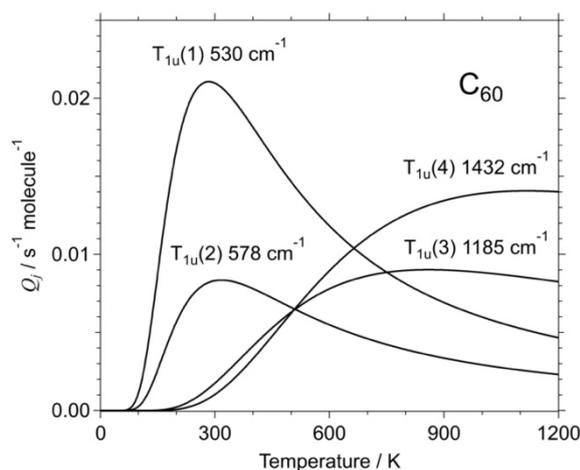


Fig. 2. Simulated temperature dependence of the IR emission intensities for the T_{1u} modes of C₆₀.

¹ Cami, J.; Bernard-Salas, J.; Peebles, E.; Malek, S. E. Detection of C₆₀ and C₇₀ in a young planetary nebula. *Science*, **2010**, *329*, 1180.

² Wakabayashi, T.; Momose, T.; Fajardo, M. E. Matrix isolation spectroscopy and spectral simulations of isotopically substituted C₆₀ molecules. *J. Chem. Phys.* **2019**, *151*, 234301.

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