

UV photochemistry and thermal processing of ketene under interstellar conditions: from rare gas matrix isolation to interstellar ice analogs

Mohamad Ibrahim¹, Jean-Claude Guillemin² and Lahouari Krim¹

¹*Sorbonne Université, CNRS, De la Molécule aux Nano-Objets: Réactivité, Interactions, Spectroscopies, MONARIS, 75005, Paris, France.*

²*Univ Rennes, Ecole Nationale Supérieure de Chimie de Rennes, CNRS, ISCR – UMR6226, F-35000 Rennes, France.*

Interstellar ketene was detected in space almost fifty years ago, back in 1977, in the dense molecular cloud Sagittarius B2. Such a simple organic species of astrophysical interest plays an important role in prebiotic astrochemistry and has been a subject of several experimental and theoretical investigations. By combining laboratory astrochemical studies in bulk amorphous ice phase and in rare-gas matrices (Ar, Ne), this work aims to understand the UV photochemistry of CH₂CO in the solid phase under interstellar conditions. The thermal processing and the UV light treatment of CH₂CO in the diluted and the concentrated phases have been monitored by means of Fourier transform infrared (FTIR) spectroscopy and mass spectrometry. Different behaviours are observed for the photolysis of ketene depending on the environment where CH₂CO is trapped. We show that while the irradiation of CH₂CO/Ne or CH₂CO/Ar samples in the 3 - 10 K temperature range leads to stable molecules (CH₄, H₂CO, C₂H₂ and CH₃CHO) and reactive radical species such as C₂O, HCCO and C₃O which are stabilized in neutral rare-gas crystals, the UV processing of CH₂CO ice analogs leads mainly to an organic residue which remains in solid phase even at room temperature. Different reaction mechanisms will be discussed to describe the evolution of the photoproducts formed during the photolysis of CH₂CO, from the bulk phase to the solid-gas interface.

Keywords : astrochemistry - molecular processes - methods: laboratory: solid state - techniques: spectroscopic - ISM: molecules